

## Theoretical Problems

### Problem 1: Ascending moist air

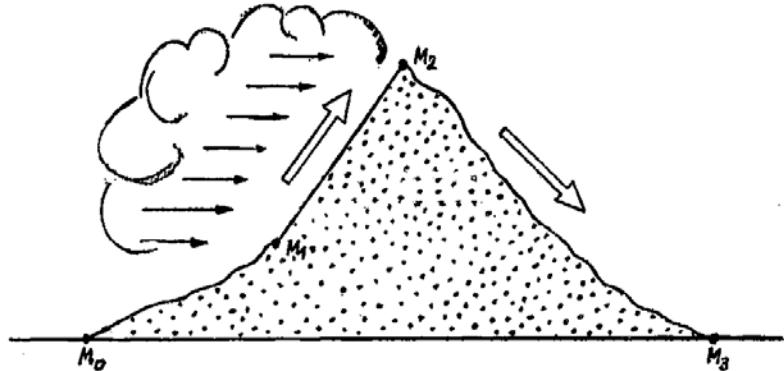
Moist air is streaming adiabatically across a mountain range as indicated in the figure.

Equal atmospheric pressures of 100 kPa are measured at meteorological stations  $M_0$  and  $M_3$  and a pressure of 70 kPa at station  $M_2$ . The temperature of the air at  $M_0$  is 20° C.

As the air is ascending, cloud formation sets in at 84.5 kPa.

Consider a quantity of moist air ascending the mountain with a mass of 2000 kg over each square meter. This moist air reaches the mountain ridge (station  $M_2$ ) after 1500 seconds.

During that rise an amount of 2.45 g of water per kilogram of air is precipitated as rain.



1. Determine temperature  $T_1$  at  $M_1$  where the cloud ceiling forms.
2. What is the height  $h_1$  (at  $M_1$ ) above station  $M_0$  of the cloud ceiling assuming a linear decrease of atmospheric density?
3. What temperature  $T_2$  is measured at the ridge of the mountain range?
4. Determine the height of the water column (precipitation level) precipitated by the air stream in 3 hours, assuming a homogeneous rainfall between points  $M_1$  and  $M_2$ .

5. What temperature  $T_3$  is measured in the back of the mountain range at station  $M_3$ ?

Discuss the state of the atmosphere at station  $M_3$  in comparison with that at station  $M_0$ .

### Hints and Data

The atmosphere is to be dealt with as an ideal gas. Influences of the water vapour on the specific heat capacity and the atmospheric density are to be neglected; the same applies to the temperature dependence of the specific latent heat of vaporisation. The temperatures are to be determined to an accuracy of 1 K, the height of the cloud ceiling to an accuracy of 10 m and the precipitation level to an accuracy of 1 mm.

Specific heat capacity of the atmosphere in the pertaining temperature range:

$$c_p = 1005 \text{ J} \cdot \text{kg}^{-1} \cdot \text{K}^{-1}$$

Atmospheric density for  $p_0$  and  $T_0$  at station  $M_0$ :  $\rho_0 = 1.189 \text{ kg} \cdot \text{m}^{-3}$

Specific latent heat of vaporisation of the water within the volume of the cloud:

$$L_v = 2500 \text{ kJ} \cdot \text{kg}^{-1}$$

$$\frac{c_p}{c_v} = \gamma = 1.4 \quad \text{and} \quad g = 9.81 \text{ m} \cdot \text{s}^{-2}$$